# I ENNTECH

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AMBERLITE IR122 Na is a premium quality, gel type, synthetic, high-capacity cation exchange resin of the sulfonated polystyrene type, supplied as small, insoluble beads in the sodium form. Generally similar to AMBERLITE IR120 Na in physical and chemicals characteristics, AMBERLITE IR122 Na has a higher level of crosslinkage and is more resistant to chemical attack when treating aggressive waters.

AMBERLITE® IRI22 Na

Strong Acid Cation Exchanger

## **PROPERTIES**

Matrix Functional Groups \_\_\_\_\_ Physical Form \_\_\_\_\_ Ionic Form, as shipped \_\_\_\_\_ Total Exchange Capacity \_\_\_\_\_ Moisture Holding Capacity \_\_\_\_\_ Shipping Weight\_\_\_\_\_ Harmonic mean size \_\_\_\_\_ Uniformity coefficient\_\_\_\_\_ Screen Grading (wet) \_\_\_\_\_ Screen Analysis

Styrene divinylbenzene copolymer Sulfonic Acid Amber beads Sodium  $2.1 \text{ meq/ml minimum (Na}^+ \text{ form)}$  $38 \text{ to } 44\% \text{ (Na}^+ \text{ form)}$  $54 \text{ lbs/ft}^3$ 0.60 to 0.70 mm 1.7 maximum 16 to 50 mesh (US Std Screens) 5 % maximum on 16 mesh (US Std Screens) 1% maximum thru 50 mesh (US Std Screens)  $Na^+ \rightarrow H^+$ : approximately 10%

Test methods are available on request.

# SUGGESTED OPERATING CONDITIONS

Maximum Reversible Swelling

pH range
Maximum operating tempeature
Minimum bed depth
Service flow rate
Regenerants (100 % basis)
Flow rate $(\text{gpm}/\text{ft}^3)$
Concentration (%)
Level (lbs/ft <sup>3</sup> )
Minimum contact time
Rinse flow rate
Rinse Requirements

250 °F		
24 inches		
$2 \text{ gpm/ft}^3$		
HCl	H <sub>2</sub> SO <sub>4</sub>	NaCl
0.3 to 0.75	0.5 to 1.0	0.5 to 1.0
4 to 10	1 to 5	10
2 to 8	5 to 15	5 to 25
30 minutes		
$1 \text{ ornm} / \text{ft}^3 \text{ in}^3$	itially then 1	$5 \text{ gal/ft}^3$

0 to 14

 $1 \text{ gpm/ft}^3$  initially, then  $1.5 \text{ gal/ft}^3$ 25 to 75  $gal/ft^3$ 

### PERFORMANCE

The operating capacity depends on several factors, such as the water analysis and the level of regeneration. The data to calculate the operating capacity and the ionic leakage with co-flow regeneration are given in the AMBERLITE IR122 Na Engineering Data Sheets.

#### LIMITS OF USE

AMBERLITE IR122 Na is suitable for industrial used. For other specific applications such as *pharmaceutical, food processing or potable water applications,* it is recommended that all potential users seek advice from Rohm and Haas Company in

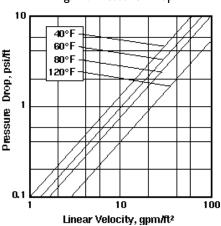


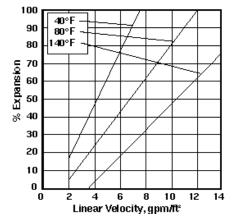
Fig. I : Pressure Drop

order to determine the best resin choice and optimum operating conditions.

#### HYDRAULIC CHARACTERISTICS

Figure 1 shows the pressure drop data for AMBERLITE IR122 Na, as a function of service flow rate and water temperature. Pressure drop data are valid at the start of the service run with clear water and a correctly classified bed. Figure 2 shows the bed expansion of AMBERLITE IR122 Na, as a function of backwash flow rate and water temperature.





# ROHM HAAS

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Ion exchange resins and polymeric adsorbents, as produced, contain by-products resulting from the manufacturing process. The user must determine the extent to which organic by-products must be removed for any particular use and establish techniques to assure that the appropriate level of purity is achieved for that use. The user must ensure compliance with all prudent safety standards and regulatory regulirements governing the application. Except where specifically otherwise stated, Rohm and Haas Company does not recommend its ion exchange resins or polymeric adsorbents, as supplied, as being suitable or appropriately pure for any particular use. Consult your Rohm and Haas technical representative for further information. Acidic and basic regenerant solutions are corrosive and should be handled in a manner that will prevent eye and skin contact. Nitric acid and other strong oxidising agents can cause explosive type reactions when mixed with Ion Exchange resins. Proper design of process equipment to prevent rapid buildup of pressure is necessary if use of an oxidising agent such as nitric acid is contemplated. Before using strong oxidising agents in contact with Ion Exchange Resins, consult sources knowledgeable in the handling of these materials.

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